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## THE MERCK INDEX

AN ENCYCLOPEDIA OF CHEMICALS, DRUGS, AND BIOLOGICALS

**ELEVENTH EDITION** 

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a-Form, monohydrate, needles, mp 159°.  $[\alpha]_0^{\infty}$  ÷98.0° → 50.0° (water). Soluble in water: slightly sol in hot alcohol. Practically insol in ether.

ractically insol in editer.  $\beta$ -Form, mp 166°. [a]<sub>0</sub> +27°  $\rightarrow$  +55.6° (water). Phenylhydrazone, mp 141°.

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4243. Galanga. Galangal; colic root; East India root; Chinese ginger. Dried rhizome of Alpinia officinarum Hance. Zingiberaceae. Habit. China. Constit. Volatile oil, resin, kaemplerid, galangin, dioxyflavanol, galangol.

4244. Galangin. 3,5,7-Trihydroxy-2-phenyl-4H-1-ben-2444. Guiungin. 3,3,7-1rinyaroxy-2-pnenyi-41-1-0en-zopyran-4-one; 3,5,7-trihydroxyflavone; norizalpinin. Cis-H<sub>10</sub>O<sub>3</sub>, mol wt 270.25. C 66.67%, H 3.73%, O 29.60%. Isoln from galanga root. Alpininia officinarum, Hance and characfrom galanga root, Alpininia officinarum, Hance and characterization: E. Jahns, Ber. 14, 2807 (1881). Prepn: T. Heap, R. Robinson, J. Chem. Soc. 129, 2336 (1926); J. J. Chavan, R. Robinson, ibid. 1933, 368. Mutagenicity studies: J. T. MacGregor, L. Jurd, Mutat. Res. 54, 297 (1978); J. P. Brown, P. S. Dietrich, ibid. 66, 223 (1979).

Yellowish needles from ethanol, mp 214-215°. Moderately sol in ethanol, ether; insol in water. Very sol in chloro-

4245. Galanthamine. 4a,5,9,10,11,12-Hexahydro-3-methoxy-11-methyl-6H-benzofuro[3a,3,2-efj[2]benzazepin-6-ol; galantamine: lycoremine; Jilkon. C<sub>17</sub>H<sub>11</sub>NO<sub>3</sub>; mol wi 287.35. C 71.05%, H 7.37%, N 4.87%, O 16.70%. From Caucasian snowdrops. Galanthus woronowii Vel.. Amaryllidaceae: Proskurnina, Yakovieva, J. Gen. Chem. 22, 1899 (1952); from Narcissus spp: Boit et al.. Ber. 90, 725, 2197 (1957). Structure work: Kobayashi et al.. Chem. & Ind. (London) 1956, 177. Synthesis and stereochemistry: Barton, Kirby, Proc. Chem. Soc. 1960, 392; J. Chem. Soc. 1962, 806; Williams. Rogers, Proc. Chem. Soc. 1964, 357. Alternate total synthesis: Kametani et al., J. Chem. Soc. (C) 1971, 1043. Asymmetric synthesis of (+)- and (-)-forms from L-tyro-4245. Galanthamine. 4a,5,9,10,11,12-Hexahydro-3-Asymmetric synthesis of (+)- and (-)-forms from L-tyrosine: K. Shimizu et al., Heterocycles 8, 277 (1977). Biosynthesis studies: D. H. R. Barton et al., J. Chem. Soc. 1963, 4545; W. Döbke, Heterocycles 6, 551 (1977).

Crystals from benzene, mp 126-127. [ $\alpha$ ] $^{18}_{10}$  – 118.8° (c = 1.378 in ethanol). Monoacidic base. Fairly sol in hot waters freely sol in alcohol, acetone, chloroform. Less sol in ben-

zene, ether. Hydrochloride, C<sub>17</sub>H<sub>21</sub>NO<sub>2</sub>.HCl, crystals from water, dec 256-257. Sparingly sol in cold, more sol in hot water. Very sparingly sol in alcohol, acetone.

Hydrobromide, C11H12BrNO3, Nivalin. Crystals from

water. dec 246-247°. [ $\alpha$ ] $_{D}^{30}$  =93.1° (c = 0.1015 in 15 ml H<sub>2</sub>O). LD<sub>50</sub> i.v. orally in mice: 8.0, 18.7 mg/kg. Umarova et al.. C.4. 66, 53993v (1967). THERAP CAT: Cholinesterase inhibitor.

4249

4246. Galegine. (3-Methyl-2-butenyl)guanidine; N-3.3-dimethylallylguanidine; isoamyleneguanidine. C<sub>6</sub>H<sub>13</sub>N<sub>3</sub>; mol wt 127.18. C 56.66%, H 10.30%, N 33.04%. Isoprenoid guanidine deriv from seeds of Galega officinalis L., Leguminosae: Tanret, Compt. Rend. 158, 1182, 1426 (1914); 159, 108 (1914); Markovic, Dittertová, Chem. Zvestí 9, 576 (1955), C.A. 50, 8137d (1956). Structure: Barger, White, Biochem. J. 17, 827 (1923). Synthesis: Späth, Spitzy, Ber. 88, 2273 (1925); Babor, Jezo, Chem. Zvestí 8, 18 (1954), C.A. 49, 7495f (1955). Metabolic effects: G. Weitzel et al., Z. Physiol. Chem. 353, 535 (1972). Effects on mitochondria: B. Lotina et al., Arch. Biochem. Biophys. 159, 520 (1973). Biosynthetic study: J. Steiniger, G. Reuter, Biochem. Physiol. Pflanz. 166, 275 (1974). Review: Braun, J. Chem. Ed. 8, 2175 (1931). 4246. Galegine. (3-Methyl-2-butenyl)guanidine; N-3.3-

Hygroscopic, bitter crystals. mp 60-65°. Freely sol in water or alcohol, slightly in ether. Keep well closed.

4247. Galipine. 2-[2-(3,4-Dimethoxyphenyl)ethyl]-4-methoxyquinoline. C<sub>20</sub>H<sub>31</sub>NO<sub>3</sub>; mol wt 323.38. C 74.28%, H 6.55%, N 4.33%, O 14.84%. From Angostura bark (Cusparia trifoliata Engl., Rutaccae): Körner, Böhringer, Gazz. Chim. Ital. 13, 363 (1883); Tröger, Kroseberg, Arch. Pharm. 250, 1494 (1912). Synthesis: Späth, Eberstaller, Ber. 57, 1687 (1924); Späth, Pikl, Ber. 62, 2244 (1929); Schläger, Leeb, Monatsh. 81, 714 (1950).

$$\underbrace{\text{OCH}_3}^{\text{N}} \underbrace{\text{CH}_2\text{CH}_2}^{\text{CH}_2} \underbrace{\text{OCH}_3}^{\text{OCH}_3}$$

Prismatic needles from alc, mp 116°. Soluble in alcohol, benzene, chloroform, ether; slightly sol in water, petr ether. The salts are more sol than those of cusparine. Hydrochloride tetrahydrate, C<sub>20</sub>H<sub>21</sub>NO<sub>3</sub>.HCl.4H<sub>2</sub>O, plates, become anhydr at 100°, mp 165°. Methiodide, C<sub>20</sub>H<sub>21</sub>NO<sub>3</sub>.CH<sub>3</sub>I, yellow needles, mp 146°.

4248. Gallacetophenone. 1-(2,3,4-Trihydroxyphenyl)eth 4248. Gallacetophenone. 1-(2,3,4-Trihydroxyphenyl)ethanone; 2',3',4'-trihydroxyacetophenone; Alizarine yellow C; C.I. 57000. C<sub>4</sub>H<sub>4</sub>O<sub>4</sub>: mol wt 168.14. C 57.14%, H 4.79%, O 38.06%. Prepn: Hart, Woodruff, J. Am. Chem. Soc. 58, 1957 (1936); Campbell. Coppinger, U.S. pat. 2,686,123 (1954 to U.S. Secy. Agr.); Knowles, U.S. pat. 2,763,691 (1956 to Kodak); Price, Israelstam J. Org. Chem. 29, 2800 (1964)

White to brownish-gray, cryst powder, mp 173°, uv max (methanol): 237, 296 nm (ε 8560, 12,500). Sol in 600 parts cold water, more in hot water; sol in alcohol, ether, soln of sodium acetate.

USE: Antiseptic.

4249. Gallamine Triethiodide. 2,2',2"-[1,2,3-Benzenetriyltris(oxy)]tris[N,N,N-triethylethanaminium] triiodide; -phenenyltris(oxyethylene)]tris[triethylammonium trilodidel; 1.2.3-tris(2-triethylammonium ethoxy)benzene triiodide: 1,2,3-tris(2-diethylaminoethoxy)benzene tris(ethyl iodide); tri(\$-diethylaminoethoxy)-1.2.3-benzene triiodom': acetone 43.5, benzene 28.9, CHCl<sub>3</sub> 24.0, ether 20.8, ethanol 6.4. Insol in water. LD<sub>50</sub> in male, female rats (mg/kg): 88, 91 orally (Gaines).

ing (Sg).

Human Toxicity: Poisoning may occur by ingestion, inhalation, or percutaneous absorption. Acute: Dizziness. headache, nausea, vomiting, diarrhea, tremors, weakness, headaches dyspnea, cyanosis, circulatory collapse. Chron-convulsions dyspnea, cyanosis, circulatory collapse. Chron-he: Hepatic damage has occurred in exptl animals. Topical Hepatic damage has occurred in expit animals. Topical use may cause local sensitivity reactions. Vapors may irritate eyes, nose, throat. See: Clinical Toxicology of Commercian Products. R. E. Gosselin et al., Eds. (Williams & Williams, Baltimore, 5th ed., 1984) Section III. pp 239-241. Lindage and other hexachlorocyclohexane isomers may remarkly be anticipated to be carringages. Fourth Aparts casionably be anticipated to be carcinogens: Fourth Annual Report on Carcinogens (NTP 85-002, 1985) p 123.

(SE: Insecticide. IRERAP CAT: Pediculicide; scabicide. IHERAP CAT (VET): Ectoparasiticide.

5380. Lindlar Catalyst. Pd-Pb-CaCO<sub>3</sub>. Prepn: Lindlar. Relv. Chim. Acta 35, 446 (1952); Lindlar, Dubuis cited by Fieser, Fieser, Reagents for Organic Synthesis (New York,

[967] p 566.
TSF: In selective hydrogenation of triple bonds to cisdouble bonds.

5381. Lineatin. [1R-(1α, 2β, 5α, 7β)]-3, 3, 7-Trimethyl-4, 9-dioxatricyclo[3, 3, 1, 0<sup>2,7</sup>]nonane; 3, 3, 7-trimethyl-2, 9-di-4,0-dioxatricyclo[3.3.1.0<sup>1.7</sup>]nonane: 3,3.7-trimethyl-2,9-dioxatricyclo[3.3.1.0<sup>4.7</sup>]nonane: 4,6,6-lineatin. C<sub>10</sub>H<sub>16</sub>O<sub>3</sub>; mol wt 168.24. C 71.39%, H 9.59%, O 19.02%. Isoln of the unique tricyclic aggregation pheromone from ambrosia heetles. Trypodendron lineatum (Olivier): J. G. MacConnell et al., J. Chem. Ecol. 3, 549 (1977). Synthesis of (±)-form: K. Mori. M. Sasuki. Tetrahedron Letters 1979, 1329; K. N. Slessor et al., J. Org. Chem. 45, 2290 (1980): K. Mori et al., Tetrahedron Letters 23, 1921 (1982): L. Skattebol. Y. Stenstrom. ibid. 24, 3021 (1983): B. D. Johnston et al., J. Org. Chem. 50, 114 (1985). Synthesis of racemate and optical synthesis. isomers: K. Mori, M. Sasaki, Tetrahedron 36, 2197 (1980). Short stereoselective synthesis: I. Aljancic-Solaja et al., Helv. Clim. Acta 70, 1302 (1987). Comparative activity of the isomers: J. H. Borden et al., Can. Entomol. 112, 107 (1980).

Oil.  $bp_{10}$  70°.  $[\alpha]_{10}^{20}$  +66.3° (c = 3.1 in CHCl<sub>3</sub>). USE: Insect sex attractant.

5382. Linoleic Acid. (Z,Z)-9.12-Octadecadienoic acid; 9.12-linoleic acid: linolic acid. C<sub>18</sub>H<sub>32</sub>O<sub>2</sub>; mol wt 280.44. C 77.09% H 11.50% O 11.41%. An essential fatty acid, q.v. Major constituent of many vegetable oils, e.g., cottonseed, suybean, peanut, corn, sunflower seed, sufflower, poppy seed, lineard, and recilla oils, where it occurs as a glyceride. soybean, peanut, corn, sunflower seed, safflower, poppy seed, linseed, and perilla oils, where it occurs as a glyceride. Characteristic ingredient of semi-drying oils. Isoln: Swern, Parker, J. Am. Oil Chem. Soc. 30, 5 (1953); Parker et al., Biochem. Frepn. 4, 86 (1955); McCutcheon, Org. Syn. coll, vol. III, 526 (1955). Summary of work on structure: T. P. Hidhich. The Chemical Constitution of Natural Fais (Chapman & Hall, London, 2nd ed. 1956). Synthesis: Raphael, Sondheimer, J. Chem. Soc. 1950, 2102; Gensler, Thomas, J. Am. Chem. Soc. 73, 4601 (1951); Walborsky et al., ibid. 2590; Nigam, Weedon, J. Chem. Soc. 1956, 4052; Osbond, Wickens, Chem. & Ind. (London) 1959, 1288. Review of physiological role in mammals: H. S. Hansen, Trends Biochem. Sci. 11, 263 (1986).

Colorless oil. Easily oxidized by air, cannot be distilled without decompn. Storage in ester form is recommended.  $\mathbf{d}_{1}^{18}$  0.9038;  $\mathbf{d}_{2}^{17}$  0.9007. mp -12°. bp<sub>14</sub> 202°; bp<sub>16</sub> 230°.  $n_{1}^{11.5}$  1.4715;  $n_{1}^{10}$  1.4699;  $n_{1}^{11.5}$  1.4683;  $n_{2}^{10}$  1.4588. lodine value: 181.1. Thiocyanogen value 96.7. Freely sol in ether. Sol in abs alc. One ml dissolves in 10 ml petr ether. Miscible with dimethylformamide, fat solvents, oils.

Aluminum salt, Al(C<sub>18</sub>H<sub>29</sub>O<sub>2</sub>)<sub>3</sub>. Yellow lumps or powder; linseed oil odor. Practically insol in water. Sol in oils, fixed alkali hydroxides.

Methyl Ester see Methyl Linoleate.

Ethyl Ester see Ethyl Linoleate. Cyclohexylamide, C<sub>2</sub>H<sub>43</sub>NO, linolexamide, N-cyclohexyl-

linoleamide, Clinolamide. USE: Manuf paints, coatings, emulsifiers, vitamins. Aluminum salt used to manuf lacquers

THERAP CAT: Nutrient (essential fatty acid)

5383. Linolenic Acid. (Z,Z,Z)-9,12,15-Octadecatricnoic acid; α-linolenic acid. C<sub>18</sub>H<sub>19</sub>O<sub>2</sub>; mol wt 278.42. C 77.65%, H 10.86%, O 11.49%. An essential fatty acid, q.v. Occurs as the glyceride in most drying oils. Synthesis: Nigam, Weedon, J. Chem. Soc. 1956, 4049; Osbond, Wickens, Chem. & Ind. (London) 1959, 1288. Biosynthetic studies: C. G. Kannangara et al., Biochem. Biophys. Res. Commun. 52, 648 (1973); B. S. Jacobson et al., ibid. 1190; C. J. Bedord et al., Arch. Biochem. Biophys. 185, 15 (1978). Effects on lipid metabolism in rat tissue: M. L. Garg et al., Lipids 23, 847 (1988). Review of dietary linolenic acid in mammals: J. Tinoco et al., ibid. 14, 166-171 (1979); in man: N. Zöllner, Prog. Lipid Res. 25, 177-180 (1986). 5383. Linolenic Acid. (Z,Z,Z)-9,12,15-Octadecatricnoic

$$H_3C$$
 $(CH_2)_7COOH$ 

Colorless liquid. dis 0.914. bp, 230-232°. Insol in water. THERAP CAT: Nutrient (essential fatty acid).

5384.  $\gamma$ -Linolenic Acid. (Z,Z,Z)-6,9,12-Octadecatrienoic acid; cis-6.cis-9.cis-12-octadecatrienoic acid; gamolenic acid; GLA. C<sub>18</sub>H<sub>20</sub>O<sub>2</sub>; mol wt 278.44. C 77.65%, H 10.86%, O 11.49%. Polyunsaturated fatty acid produced in the body as the Δ<sup>6</sup>-desaturase metabolite of lineleic acid, q.v. Converted to dihomo-y-linolenic acid, a biosynthetic precursor of monoenoic prostaglandins such as PGE<sub>1</sub>. Present to varying extents in the farty acid fraction of evening primrose oil (7-10%), in borage oil (18-26%), in black currant oil (15-20%) and in oils from different fungal sources (6-24%). 20%) and in oils from different tungal sources (6-24%). Isoln from evening primrose oil, q.v.: A. Heiduschka, K. Luft, Arch. Pharm. 257, 33 (1919). Proposed structure: Eibner et al., Chem. Umschau. 34, 312 (1927). Confirmation of structure: J. P. Riley, J. Chem. Soc. 1949, 2728. Discussion of occurrence, esp. in fungi: R. Shaw, Biochim. Biophys. Acta 98, 230 (1965). Synthesis: J. M. Osbond et al., J. Chem. Soc. 1961, 2779; J. M. Osbond. ibid. 5270. Metabolism studies: J. F. Mead, D. R. Howton, J. Biol. Chem. 229, 575 (1957); K. J. Stone et al., Linick 14, 174 (1979). Effect 575 (1957); K. J. Stone et al., Lipids 14, 174 (1979). Effect of source on essential fatty acid and prostanoid metabolite formation: D. K. Jenkins et al., Med. Sci. Res. 16, 525

Hexabromide deriv.  $C_{18}H_{30}Br_6O_2$ , crystals from ethyl methyl ketone, mp 201-202°. USE: Nutrient.

THERAP CAT: In treatment of atopic eczema.

5385. Linseed, Flaxseed; linum. Dried ripe seeds of num usitatissimum L., Linaceae. Source of linseed oil. Linum usitatissimum L., Linaceae. Source of linseed oil. Constit. 30-40% oil, about 6% mucilage, about 25% proteins and linamarin.

USE: Emollient.
THERAP CAT (VET): Poultice (crushed seeds), demulcent (boiled in water).

5386. Linseed Oil, A drying oil obtained by expression of linseed. Constit. Glycerides of linolenic, linoleic, oleic, stearic, palmitic and myristic acids. Ref: T. P. Hilditch. The Chemical Constitution of Natural Fats (London. 3rd ed..